

Oxygen Containing Organic Compounds-II

Date Planned ://	CBSE Pattern	Expected Duration : 90 Min
Actual Date of Attempt : / /	Level-0	Exact Duration :

Very Short Answer Type

(1 Mark)

- 1. IUPAC name of the following compound Ph CH = CH CHO.
- **2.** Write the product in the following reaction :

$$\mathrm{CH_{3}-CH}=\mathrm{CH-CH_{2}CN}\xrightarrow{\quad \text{(i) DIBAL-H} \quad } \mathrm{(A)}$$

- 3. Aldehydes and ketones have lower boiling points than corresponding alcohols why?
- **4.** Distinguish between the following using suitable chemical test.

(i)
$$\begin{array}{ccc} O & O \\ || & || \\ C_6H_5-C-CH_3 \text{ and } C_6H_5-C-H_3 \end{array}$$

5. Name the reagent used in the following reaction :

$$\mathrm{C_6H_5CH_2CH_3} \xrightarrow{\quad ? \quad} \mathrm{C_6H_5COO^-K^+}$$

- **6.** Give reason chloral hydrate is a gem diol but stable.
- **7.** Write chemical equation to illustrate the following name reaction: Cannizzaro reaction

Short Answer Type-I (2 Marks)

- **8.** Write equation involved in the following reaction :
 - (i) Stephens reaction

- (ii) Etard reaction
- 9. An organic compound A, having the formula C_3H_8O , on heating with copper at 573 K, gives B. B doesn't reduce Fehling's solution but gives a yellow ppt. of the compound C with I_2 /NaOH. Deduce the structure of A, B and C?
- **10.** Write the structure of A and B in the following reaction :

$$\mathrm{CH_3COCl} \xrightarrow{\quad \mathrm{H_2}, \; \mathrm{Pd-BaSO_4} \quad} \mathrm{A} \xrightarrow{\quad \mathrm{H^+} \quad} \mathrm{B}$$

- **11.** Describe the following reaction?
 - (i) Acetylation

- (ii) Aldol condensation
- **12.** Predict the product of the following reaction :

(i)
$$CH_3 - C - CH_3 \xrightarrow{\text{(i) NH}_2 \text{NH}_2} A$$

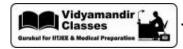
$$C_6H_5 - C - CH_3 \xrightarrow{\text{NaOH}/I_2} B + C$$

13. Predict the product of the following reaction :

(i)
$$\begin{array}{c} O \\ C - CH_3 \\ \hline \\ H_2CrO_4 \end{array}$$

$$\begin{array}{c} O \\ \parallel \\ C-Cl \\ \hline \\ Pd/BaSO_4 \end{array}$$

(ii)



14. (a) Write the product of the following reaction :

$$\text{CH}_3\text{CHO} \xrightarrow{\text{HO}^-/\Delta}$$

(b) Given simple chemical tests to distinguish between the following pairs of compounds: Benzaldehyde and benzoic acid.

Short Answer Type-II (3 Marks)

- 15. Write the structure of the main products when acetone $(CH_3 C CH_3)$ reacts with the following reagents:
 - 1. Zn Hg/conc. HCl

- 2. $NH_2NHCONH_2 / H^+$
- **3.** CH_3MgBr and then H_3O^+
- **16.** How will you bring about the following conversions?
 - (i) Propanone to propane
 - (ii) Benzoyl chloride to benzaldehyde
 - (iii) Ethanal to but-2-enal
- **17.** Identify X, Y and Z in the following sequence of reactions.

(ii)
$$\xrightarrow{\text{CHO}} \xrightarrow{\text{KOH}} X \xrightarrow{\text{H}^+} Y$$

- $\textbf{18.} \qquad \text{Write the chemical equations to illustrate the following name reaction:} \\$
 - (i) Wolff kishner reduction
- (ii) Aldol condensation

- (iii) Cannizzaro reaction
- **19.** Write the structure of the main product of the following reactions.

(i) +
$$C_6H_5COC1 \xrightarrow{Anhydrous}$$

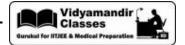
(ii)
$$CH_3 - C \equiv C - H \xrightarrow{Hg^{2+}/H_2SO_4} H_2O$$

(iii)
$$(iii) \qquad (i) \operatorname{CrO_2Cl_2} \atop (ii) \operatorname{H_2O}$$

- 20. An organic compound with molecular formula $C_9H_{10}O$ form 2, 4-DNP derivative, reduces Tollen's reagent and undergoes Cannizzaro's reaction. On vigorous oxidation it gives 1, 2-benzenecarboxylic acid. Identify the compound.
- 21. How will you convert ethanal into the following compounds? Give the chemical equation involved.

(i)
$$CH_3 - CH_3$$

(iii) CH₃CH₂OH



Long Answer Type (5 Mark)

- 22. Consider all isomeric ketones including stereoisomers having molecular mass equal to 100.
 - (a) Draw structure of ketones that undergoes haloform reaction
 - (b) Draw structure of ketones that produce racemic product on reduction by using NaBH₄.
- 23. (a) Identify A, B and C in following sequence of reaction $CH_3CHO \xrightarrow{\text{(i) } C_2H_5MgCl} A \xrightarrow{\text{conc.}} B \xrightarrow{\text{HBr}} C$
 - **(b)** Predict the structure of the products formed when benzaldehyde is treated with NaOH.
 - (i) conc. NaOH
 - (ii) HNO_3 / H_2SO_4 (at 273-283 K)
- **24.** (i) Identify A to E in the following series of reaction:

$$\begin{array}{c} \text{CH}_{3} \\ + \text{CrO}_{3} + (\text{CH}_{3}\text{CO})_{2}\text{O} \\ & \begin{array}{c} 273\text{-}283\text{K} \\ \end{array} \\ \text{A} \\ + \text{CrO}_{3} + (\text{CH}_{3}\text{CO})_{2}\text{O} \\ & \begin{array}{c} \text{A} \\ \text{H}_{3}\text{O}^{+} \\ \end{array} \\ \text{B} \\ \text{conc. NaOH} \\ \text{E} \\ \end{array}$$

(ii) $C_6H_6 \xrightarrow{CH_3COCl} Anhy AlCl_3 A \xrightarrow{Zn - Hg/conc. HCl} B$ $\downarrow NaOl \qquad \qquad \downarrow (ii) KMnO_4 KOH, \Delta (iii) H_3O^+$ C

25. (i) Identify A to E in the following reactions :

COOH
$$\begin{array}{c}
\text{conc. HNO}_{3} \\
\text{SOCl}_{2}
\end{array}$$

$$\begin{array}{c}
\text{SOCl}_{2} \\
\text{D} \\
\begin{array}{c}
\text{H}_{2}/\text{Pd/BaSO}_{4} \\
\text{guingline}
\end{array}$$

$$\begin{array}{c}
\text{E}
\end{array}$$

(ii) An organic compound (A) on treatment with acetic acid in the presence of sulphuric acid produces an ester (B), (A) on mild oxidation gives (C). (C) with 50% KOH followed by acidification with dil HCl generates (A) and (D). (D) with PCl_5 followed by reaction with ammonia gives (E). (E) on dehydration produces hydrocyanic acid.

Identify the compounds A, B, C, D and E.